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Introduction to Atoms Quiz. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Rachel\_Frishberg. Terms in this set (8) Nucleus. The very small center core of an atom. It contains the protons and neutrons. Protons. Located inside of the nucleus. Have a positive (+) electric charge.

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2) All atoms of the same element are exactly alike and have the same mass. Atoms of different elements are different and have different masses. 3) An atom of one element cannot be changed into an atom of a different element by a chemical reaction. 4) Compounds are formed when atoms of more than one element combine in a specific ratio.

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9 Name Date Class © Pearson Education, Inc., publishing as Pearson Prentice Hall. All rights reserved. \_\_\_\_ 8. According to Boyle's law, as the volume of a gas is made

## **Introduction to Matter - Bridgeway**

Name Elements and the Periodic Table Date Guided Reading and Study Class Introduction to Atoms (pp. 102-108) This section describes the development of atomic models and the structure of atoms.

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